

monitoring of the first 7 minutes of reactivity and shows essentially no heat flow between 0.0°C and approximately 20°C, at which point onset of endothermic heat flow occurs.

*C₃
Conced'*
The heat flow properties indicate that at 37°C the reaction is essentially endothermic, and under the conditions used, the reaction occurs only very slowly if at all at temperatures below about 20°C. Thus, the net reactivity in the system, that is, the sum of endothermic and exothermic heat flow of the system, is endothermic.--

In the Claims

Please amend claims 2 and 25 as follows.

C₄
2. (Three times amended) A method for treating a bone defect, comprising:
identifying a bone site suitable for receiving an implant;
introducing a hydrated precursor paste at the implant site, the hydrated precursor paste comprising an amorphous calcium phosphate and a promoter; and
[converting] hardening the hydrated precursor paste *in vivo* at the implant site
wherein the hardening process is associated with an endothermic reaction [to a hardened poorly crystalline apatitic (PCA) calcium phosphate], whereby bone is formed at the implant site.

C₅ *C₁₀₄* 25. (Amended) A method for embedding a prosthetic device, comprising:
introducing a prosthesis at an implant site;